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**and Haloarene Intext 10.2 to  
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exercise solution part-1  
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Haloarenes  
Haloarenes are less polar  
than haloalkanes and are  
insoluble in water. This is

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haloarenes  
because of lack of hydrogen bonding. As a result, the attractive forces in haloarenes–water system remain less than the attractive forces in H<sub>2</sub>O molecules which are hydrogen bonded.



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Question 10.1: Name the  
following halides according  
to IUPAC system and classify

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Haloarenes  
them as alkyl, allyl, benzyl  
( primary, secondary,  
tertiary ), vinyl or aryl  
halides :

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The bond (carbon-halogen bond) of alkyl halide is polarised. Halogen atom carries a partial negative charge Carbon atom carries a partial positive

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are classified as fluoro,

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chloro, bromo or iodo  
compounds according to the  
type of halogen present and  
as mono-, di- tri-, tetra-  
haloalkanes, etc., according  
to the one, two, three,  
four, etc., halogen atoms  
respectively present in

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Chapter 10  
Chemistry Notes for class 12  
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Haloarenes. The replacement

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of hydrogen atom(s) in hydrocarbon, aliphatic or aromatic, by halogen atom(s) results in the formation of alkyl halide (haloalkane) and aryl halide (haloarene), respectively. Classification of Halogen Derivatives.

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Haloalkanes are the halogen derivatives of hydrocarbons and these are of the following types depending



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upon the number of hydrogen atoms present in them.

Haloarenes are aromatic compounds in which the halogen atom is attached directly to the carbon atom of the aromatic ring. The solutions of chapter 10 also

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give exemplary problems,  
HOTS (Higher Order Thinking  
Skills), MCQs exercises,  
assignments and worksheets  
which will help you to learn  
the concept comprehensively  
and to make haloalkanes ...

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solve NCERT Class 12  
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preparation level.

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hydrogen atom (s) in a

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Hydrocarbons, aliphatic or aromatic, by halogen atom (s), results in the formation of alkyl halide (haloalkane) and aryl halide (haloarene), respectively. Haloalkanes contain halogen atom (s) attached to the  $sp^3$



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Haloalkanes and Haloarenes.  
Nature of C-X bond in alkyl  
halides: X is more  
electronegative than carbon.  
So, the C-X bond is



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polarized with C having a partial positive charge and X having a partial negative charge. Preparation of haloalkanes:

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Preparation of Haloalkanes

I: 2: Preparation of

Haloalkanes II: 3:

Properties of Haloalkanes I:

4: Properties of Haloalkanes

II: 5: Properties of

Haloalkanes III: 6: Allyl

Halides and Dihalides I: 7:

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Reactions Of Dihalides I: 8:  
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I: 10: Solved Problem Set I:  
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classified as fluoro,  
chloro, bromo or iodo  
compounds according to the



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Hydrocarbons  
type of halogen present and  
as mono-, di- tri-, tetra-  
haloalkanes, etc., according  
to the one, two, three,  
four, etc., halogen atoms  
respectively present in  
their molecule. 2.

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